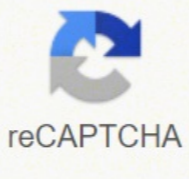




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READING

Lewis Carrol

Lewis Carrol is one of the most famous writers for children of all time. He is the author of the children's classics *Alice's Adventures in Wonderland* and *Through the Looking-Glass*.

Lewis Carrol was born on January 27, 1832 in Daresbury, Cheshire, England. His real name was Charles Lutwidge Dodgson. He was the oldest boy in a family of eleven children (seven girls and four boys). As a child, he excelled in mathematics and won many academic prizes. He also created games for himself and his brothers and sisters.

Lewis Carrol studied at Richmond School (1844–45) and Rugby School (1846–1850). In 1850, he started studying at Christ Church, Oxford. After finishing his studies, he became a teacher of mathematics at Oxford. Apart from teaching mathematics, he also took photos and wrote essays and poems.

Lewis Carrol loved children. He became friends with the children of Henry George Liddel, Alice and her sisters Lorina and Edith. The three little girls and their governess Miss Prickett visited the young mathematics teacher in his college rooms. They spent many hours with Carrol, sitting on his couch while he told them fantastic stories about dream worlds. He also drew illustrations for these stories.

During an afternoon picnic in 1864, Dodgson told the Liddel children a story of *Alice's Adventures Underground*. Alice liked it very much and when they returned home, she asked him to write it down for her. He fulfilled the small girl's wish and wrote the story down. He illustrated it with his own drawings and gave it to Alice.

One day, the story fell into the hands of the novelist Henry Kingsley. He read it and told Carrol that he should publish it. The book was published as *Alice's Adventures in Wonderland* in 1865. It became very successful. Carrol wrote some more stories and in 1871 published *Through the Looking-Glass and What Alice Found There*.

Lewis Carrol died on January 14, 1898. By the time of his death, *Alice* had become the most popular children's book in England, and by 1932 it was one of the most popular children's books in the world.

1 Read the text and answer the following questions.

- 1 How many brothers and sisters did Lewis Carrol have? _____
- 2 What did Lewis Carrol teach at Oxford? _____
- 3 What other things did he do? _____
- 4 What did Carrol do when Alice Liddel and her sisters visited him? _____
- 5 Who told Lewis Carrol that he should publish his stories about Alice? _____
- 6 What is the name of the second book about Alice? _____
- 7 When did Lewis Carrol die? _____

2 Are these sentences true (T) or false (F)?

- 1 Lewis Carrol was the oldest son of his parents. _____
- 2 Lewis Carrol was good at mathematics when he was a child. _____
- 3 Lewis Carrol studied at Richmond, Rugby and Cambridge. _____
- 4 Lewis Carrol illustrated some of the stories that he told Alice. _____
- 5 The Liddel children visited Lewis Carrol without any adult. _____
- 6 Lewis Carrol invented the stories about Alice during a picnic. _____
- 7 In 1865, Lewis Carrol published a book called *Alice's Adventures Underground*. _____

SOURCES: <https://www.biography.com/people/lewis-carroll-9239598>, <https://www.britannica.com/biography/Lewis-Carroll> ISLCollective.com

Scientific Method

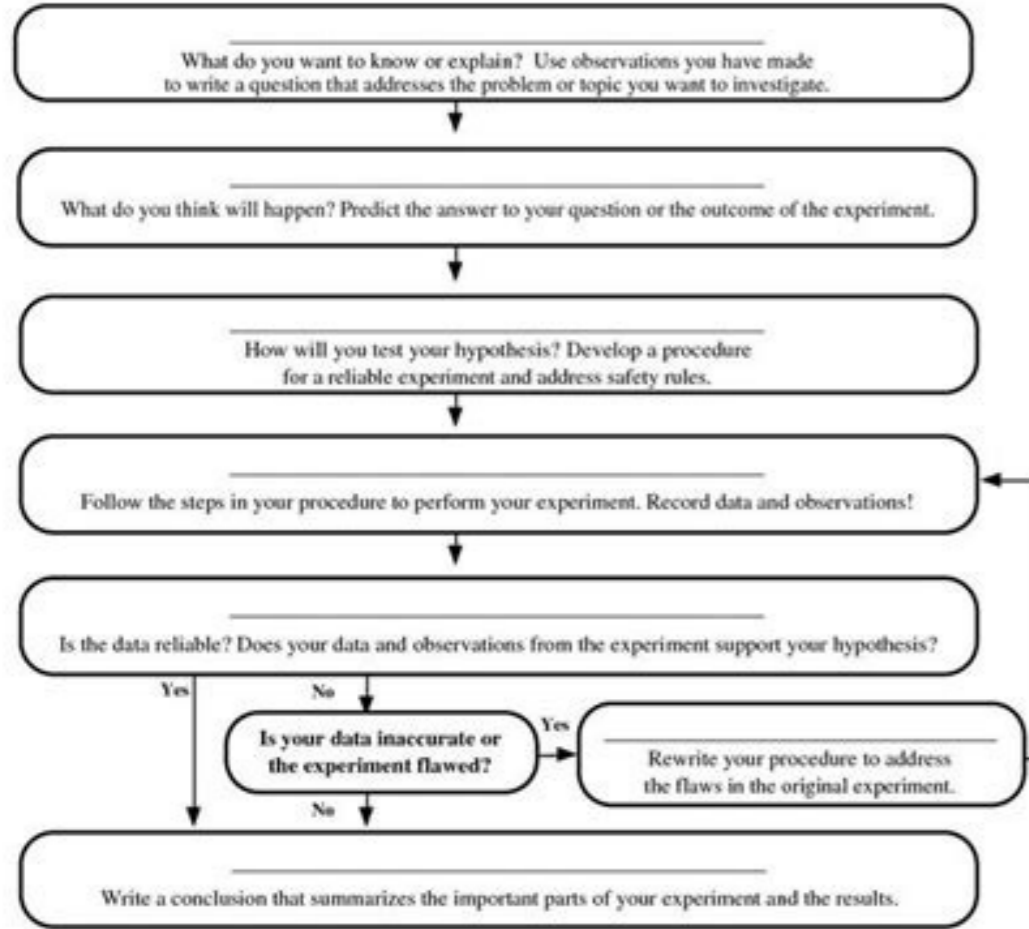
Name _____

What is the scientific method? It is a _____ that is used to find _____ to questions about the world around us.

Is there only one "scientific method"? No, there are several versions of the scientific method. Some versions have more _____, while others may have only a few. However, they all begin with the identification of a _____ or a _____ to be answered based on observations of the world around us and provide an _____ method for conducting and analyzing an experiment.

What is a hypothesis? It is an _____ based on observations and your knowledge of the topic.

What is data? It is _____ gathered during an experiment.



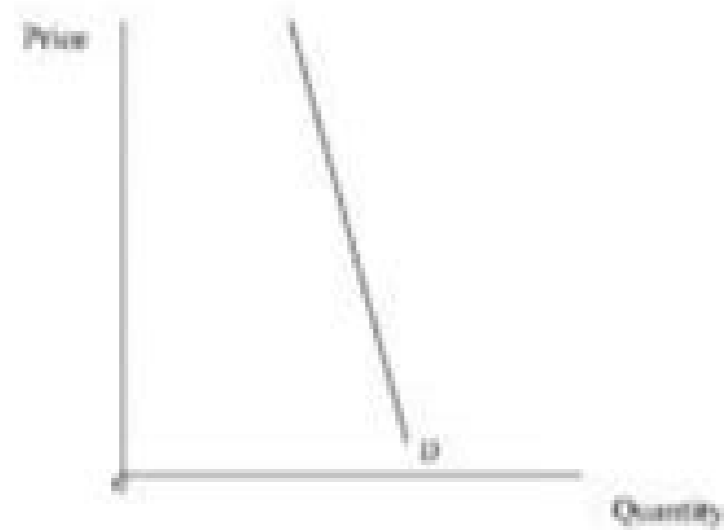
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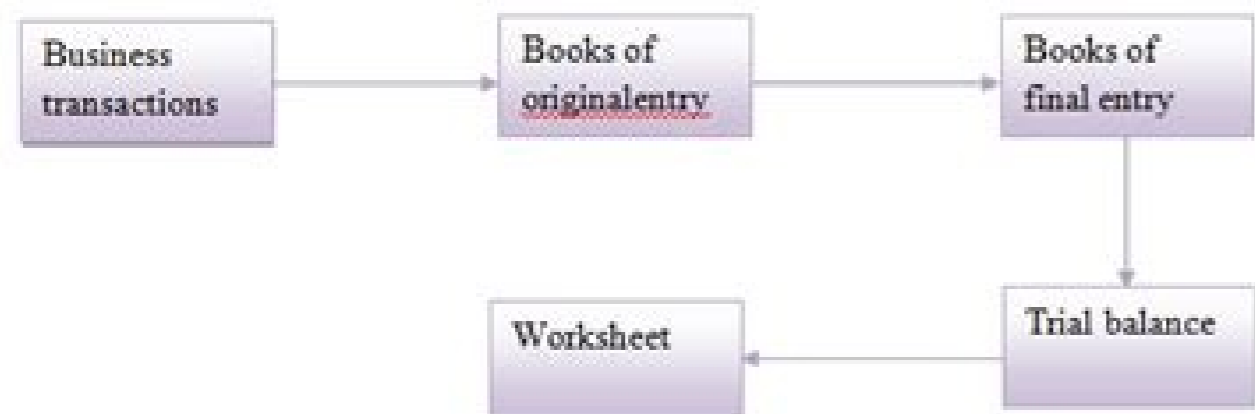
Supply and Demand Test

- Which of the following will *not* change the demand for oranges?
 - A change in consumer income
 - An expectation that the price of oranges will increase in the future
 - A change in the price of oranges
 - A change in the price of grapefruits, a substitute for oranges
- If hot dogs are an inferior good, an increase in income will result in
 - An increase in the quantity demanded for hot dogs
 - An increase in the demand for hot dogs
 - A decrease in the quantity demanded for hot dogs
 - A decrease in the demand for hot dogs
- What does the law of demand state?
 - As the price of a good increases, supply decreases
 - As the price of a good increases, demand decreases
 - As the price of a good increases, quantity demanded decreases
 - As the price of a good increases, quantity supplied decreases

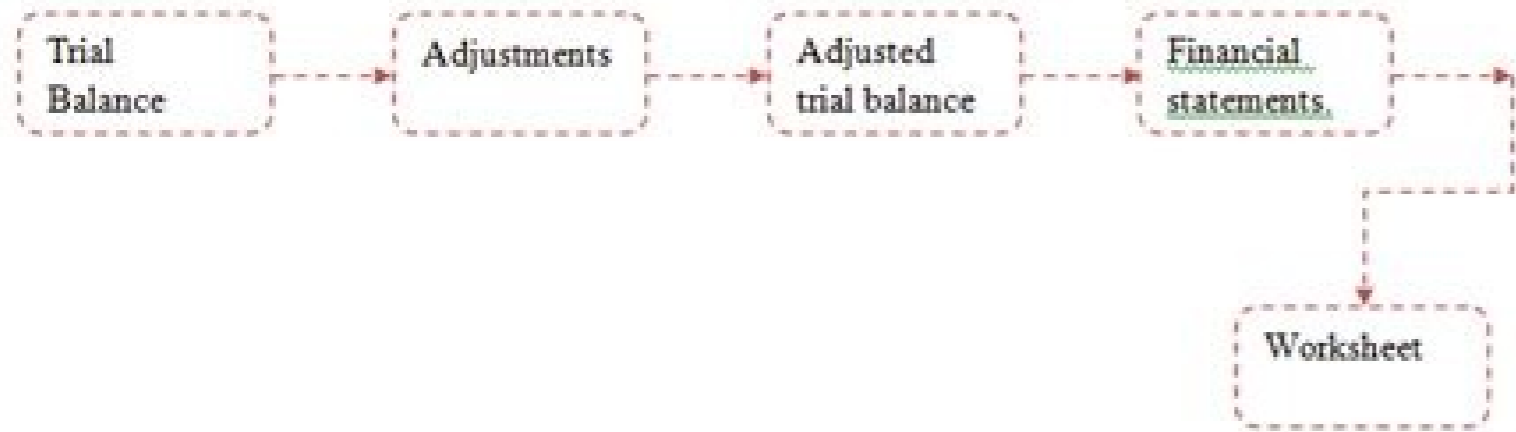


- The elasticity of the demand curve above can best be described as
 - Relatively inelastic
 - Relatively elastic
 - Perfectly inelastic
 - Perfectly elastic
- Which of the following does *not* cause a shift in supply?
 - Price of inputs
 - Technology
 - Price of complements
 - Government taxes

Teacher's Guide	ICivics
Limiting Government	
<p>Time Needed: One Class Period</p> <p>Materials Needed: Student worksheets</p> <p>Class Instructions: Reading (2 pages, class set) Graphic Organizer (2 pages, class set) Worksheets (2 pages, class set)</p>	<p>Learning Objectives: Students will be able to:</p> <ul style="list-style-type: none"> Describe the powers of government, including separation of powers, and the limits of government, and rights of the minority. Analyze the limits of government, including presidential actions affected by government limits. Explain the benefits and drawbacks of going one better than others. Apply the five links to federal government systems.
Step by Step	
<p>1. Anticipate by asking students to imagine a country ruled by one person. What powers would that person have? Can students think of any problems that might come up?</p>	
<p>2. Discuss the Limiting Government reading and the Five Links on Government graphic organizer.</p>	
<p>3. Read the first page with the class, pausing to discuss the basic concepts outlined in the reading. After each paragraph, have students complete the links on their graphic organizers.</p>	
<p>4. Read the story of George Washington with the class, stopping to check for understanding and clarify as needed.</p>	
<p>5. Check for understanding by doing the "5, 4, 3, 2, 1" Active Participation Activity with the class. (See Teacher's Guide). Watch for incorrect answers, and use each statement as an opportunity to quickly review the concept before moving on.</p>	
<p>6. Discuss the worksheets to the class.</p>	
<p>7. Review the instructions for the activities. For activity G (Guessing Game), make sure students know they can use their graphic organizers for help.</p>	
<p>8. Assess students to complete the worksheets. If you have an interactive white board, consider projecting blank masters and completing some of the activities together as a class.</p>	
<p>9. Review the answers if you wish.</p>	
<p>10. Close by asking students to quietly read the five links on government, without looking at their notes. Have them quiz a partner to read at five.</p>	



The preparation of a worksheet is usually the last step. When preparing a worksheet, the following steps have to be followed]



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Now you know how it works, try the questions to see how well you understand it. The lower temperatures would increase the performance more, but reduce the reaction rate. Our mission is to educate the next generation of successful people by producing high quality resources that improve teaching and learning. Fritz was severely criticized for his involvement in the gas war program. We cover a wide range of subjects and age groups with our resources which mainly have a scientific or mathematical focus. Ult-updated 29 November 2021a double-sided worksheet and fully complete answer sheet covering the haber process. the two had two sons named Eva-Charlotte and Ludwig-Fritz. the temperature used for the process of haber is about 450 ° c. key facts and information early life and education fritz haber was born on 9 December 1868 in breslau (which is now known as Wrocław, in polonia) in prussia. in May 1891, Fritz received his doctorate from Friedrich wilhelm university. each link takes you to a page with a general background on the topic, whether it be countries like germany and japan, whether it was like wingska and nevada. Fritz later proposed the use of chlorine gas as a chemical weapon because of the needs of germany for tear gas and other chemical irritants. during this period, Fritz was also able to publish a book on the thermodynamics of gas technical reactions. is used to produce nitrogen-based fertilizers. ammonia turns into liquid and is stored. or these printables as aid to the study, pop quiz or launch points for discussions on the various states and countries included here. add these free printable geography worksheets to your school day for home to strengthen geography skills and for variety and fun. Fritz also studied the liabilities of metals that are not rare and how the electrical current affects the corrosion of metals. fritz haber worksheets this is a fantastic bundle that includes Cie that you have to know on Fritz Haber on 22 in -depth pages. Take a look at our hundreds of other resources available in our shop. Haber's process is an example of an important important. This multiple text books of chemistry does not mention, which is that chemical factories are there to make profits, not chemical. if a company cannot make a product for less money than for what they can sell it, they will not be doing it for long. Fritz worked alongside Wilhelm Ostwald, a German chemist, and were able to integrate the process of oxidation of ammonia and nitrate synthesis and the creation of explosives. Real life is a little more complicated than that. if you refer to any of the content on this page on your website, use the following code to quote this page as an original source. Fritz Haber Facts & Worksheets: - KidsConnect, 20 August 2020Link will appear as Fritz Haber Facts & Worksheets: - KidsConnect, 20 August 2020 Usage with any resume These worksheets have been specifically designed for use with any international resume. Because the reaction is reversible, we must choose the right temperature and pressure conditions to produce ammonia. Fritz married Clara Immerwahr on 3 August 1901 and had a son named Hermann. At the University of Jena, Fritz worked as an assistant to Ludwig Knorr. It is also a useful process to think about when you are learning chemistry - because it sums up ideas on reversible reactions, reaction rates and industrial chemistry. Under the guidance of Bunte, Fritz studied the thermal decomposition of hydrocarbons and was able to process the analysis according to which the thermal stability of carbon bond is greater than the thermal stability of carbon in aromatic compounds and is smaller than in aliphatic compounds. After completing his service, Fritz returned to universityHe became a student under the guidance of Carl Liebermann. Chemical war during the First World War, worldwide, has worked to meet Germany's war needs for synthetic chemicals and alternatives. Both hydrogen and nitrogen are relatively simple to obtain - we can get nitrogen from air, and hydrogen from methane. The Haber-Bosch process is a catalytic formation of ammonia from atmospheric nitrogen and hydrogen, while under high temperature and pressure conditions. The worksheet is based on the UK curriculum content, but it can be useful for the curriculum of other countries, about 14 to 16 years students. Fritz Haber is a German chemist known for receiving the Nobel Prize for Chemistry for his work on nitrogen fixing. From 1894 to 1911, Fritz remained at the University of Karlsruhe where he succeeded in contributing to many areas. Not ready to purchase a subscription? Siegfried married again, this time in Hedwig Hamburger. Best wishes, greenAPLEmpty answer makes no sense to the end user Thanks a really useful worksheet The empty answer makes no sense to the end userThis resource to let us know if it violates our terms and conditions. So the best way to convert as much hydrogen and nitrogen into ammonia as possible is to perform the reaction to a higher and lower temperature pressure as possible. At the age of 65, Fritz died of heart failure on 29 January 1934. In 1889, Fritz left the university to do the compulsory year of voluntary service in the sixth Field Artillery Regiment. So, how do we do ammonia? Siegfried and Hedwig had three daughters, Else, Helene and Frieda. The production of ammonia in large quantities has contributed to the production of much larger quantities of fertilizers based on nitrogen, which in turn has led to a much greater agricultural yield, which has prevented billions of people from dying because of hunger. Fritz was the son ofHaber and Siegfried Haber, who were cousins that got married despite the opposition of their families. The disadvantage of this is that high pressures are expensive to make. This means that real chemical factories making ammonia use the Haber processÁ Running at about 200 atmospheres (so 200 times higher than the pressure of the atmosphere). After Clara's death, Fritz remarried to Charlotte Nathan on October 25, 1917 in Berlin. The Haber process makes an important chemical. Making a chemical works strong enough to withstand pressure higher than 200 atmospheres is not viable when making ammonia. To further build up the process of ammonia production, Fritz turned to Carl Bosch and they were able to create large-scale production of ammonia. NOBEL PRIZE IN CHEMISTRY While at the Karlsruhe, Fritz was able to invent the Haber-Bosch process alongside his assistant Robert Le Rossignol. Once the reaction has reached equilibrium, the reacting mixture is cooled down. Fritz was also able to come up with the theoretical basis for measuring electrolytic potentials and glass electrodes. The answer sheet contains the worked answers and a mark scheme, making it useful to set as an assessment task or as a piece of marked homework. Ammonia is an incredibly useful chemical. What conditions are actually used and why? After the war, Fritz got heavily involved in administrative responsibilities and in several international scientific organizations. The first lesson on dynamic equilibrium (SC12a) is also needed for Combined Science students. The bundle includes three full lesson This worksheet can be used as an independent learning tool to help students summarise the industrial conditions for the Haber process.It is designed to be used with my GCSE Chemistry lesson on Dynamic Equilibrium, which is available to buy on this site. In Bunte's institute, Fritz became a Privatdozent where he took on the role of teaching duties related to the scope of dye This means that it would take more time to make a certain amount of ammonia, and for enterprises, the costs of time of money. without these, it would be impossible to cultivate enough food to feed all the people of the world. career fritz worked for her father's dye company for a while, but later decided to pursue an academic career. in 1894 fritz became assistant to dark hans. siegfried was a popular city trader who established its activity related to dyeing pigments, paints and drugs. paula, on the other hand, had a difficult pregnancy and died three weeks after giving birth to fritz. You can use these as-is worksheets, or edit them using google slides to make them more specific to student skill levels and curriculum standards. These are ready-to-go fritz haber worksheets which are perfect for teaching students on fritz haber which was a German chemist who is known to receive the nobel premium for chemistry for his work related to nitrogen fixing. Fritz later received the Nobel Prize for Chemistry of 1918 for his work. Fritz, with the carl engler guide, managed to find the explanation of autoxidation using electrochemical terms, in which it was able to differentiate dry autoxidation. German geography Japan South America dynamic balance and haber process gcse chemistry 3 lesson bundle this package meets all the specification points needed to teach topic 15 "Dynamic Equilibria" from the chemistry course edexcel gcse 9-1 (only Triple/Separate science). in 1919 fritz proposed a graphic method to calculate energies from ionic crystals, which became known as Born-Haber cycle. Fritz and friedrich bran were able to explain the steps in the textile printing processes that adolfHe developed. During this period, Fritz also participated in conferences held by Otto Witt on the chemical technology of the dyes. Ammonia has the NH3 chemical formula. NH3, a while, Fritz went back to Berlin where he attended the Technical College of Charlottenburg, which is now known as the Technical University of Berlin. iteÁÁÁs really important that chemical factories can make things cheaply. The reaction needed to make ammonia is: N2Á Á+ 3H2Á ÁeÁÁÁ 2NH3 Remember that eÁÁÁ means a reversible reaction -Á ÁÁ Áat a molecular level, we can see both the making and breakup of ammonia all the time. In theory, they could increase the pressure even more, and increase the yield. Complete List Of Included Worksheets Fritz Haber Facts More Info Time NH3 Pride Word Process C. Green APL specialises in developing highly useful resources which can be used in the classroom and beyond. The forward reaction also reduces the volume of the gas (there are four molecules on the left hand side, and two on the right hand side), so that direction is favoured at high pressures. See the fact file below for more information on the Fritz Haber or alternatively, you can download our 22-page Fritz Haber worksheet pack to utilise within the classroom or home environment. Our customer service team will review your report and will be in touch. The forward reaction (nitrogen + hydrogen eÁÁÁ ammonia) is exothermic, so that direction is favoured at lower temperatures. Although the conditions only convert about 40% of the reactants into ammonia, the leftover hydrogen and nitrogen are not wasted. At the age of 11, Fritz attended St. Elizabeth classical school. What conditions would maximise the yield? yield?



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